

Denitrification process in energy sector and transport as source of increased nitrous oxide emissions

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Overview

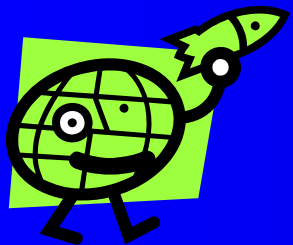
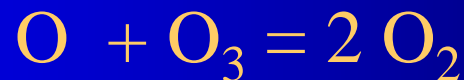
- Theoretical part
- Measurement method
- Experimental part

Coal combustion, municipal solid waste, mobile sources - cars

- Conclusion



Nitrous oxide is one of the trace gases that contribute to greenhouse warming as well as stratospheric ozone depletion.



N₂O – Greenhouse Gas

Greenhouse Gas	Global Warming Potential	Antropogenic sources (%)	Long Lifetime on Atmosphere (Years)
CO ₂	1	60	5 – 10
CH ₄	21	15	10
N ₂ O	315	6	10 - 150

*Without H₂O



The global contribution of N₂O by sectors

The share of anthropogenic emissions by sector varies according to world region and specific economic activities. Based on various studies the approximate global contribution by sector is follows:



The global contribution of N₂O by sectors

- **Agriculture and land 50-70%**
- **Industrial sources, mainly nitric acid and adipic production 15-25%**
- **Stationary and mobile fossil fuel combustion 15-25%**



Production of N_2O from fossil fuel combustion

Oxidation of the fuel-nitrogen

N_2O is formed via two different pathways

- gas phase oxidation of nitrogenous groups in the volatile – HCN,
- the heterogeneously catalysed oxidation of the char-bound N – species



Production of N_2O from fossil fuel combustion

Denitrification processes, NO and NO_2 are removed, source of N_2O

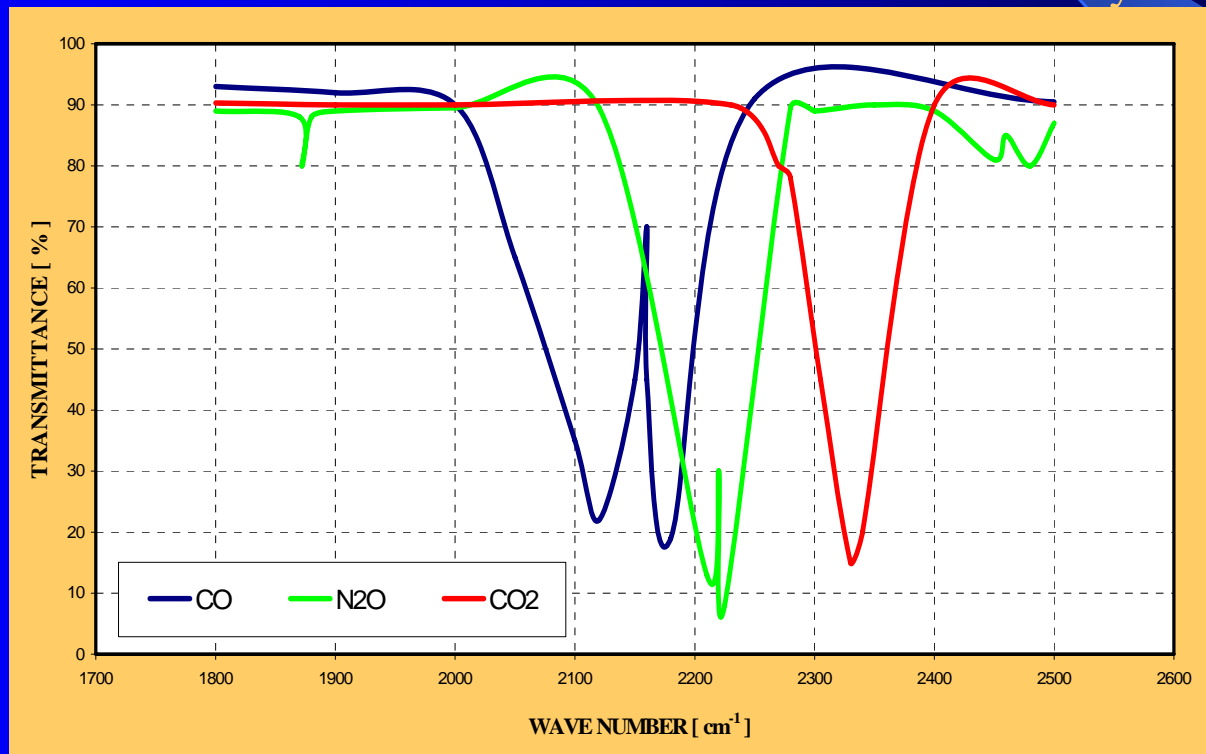
- Primary denitrification
modification of combustion process
- Secondary denitrification
post-combustion catalytic or non catalytic



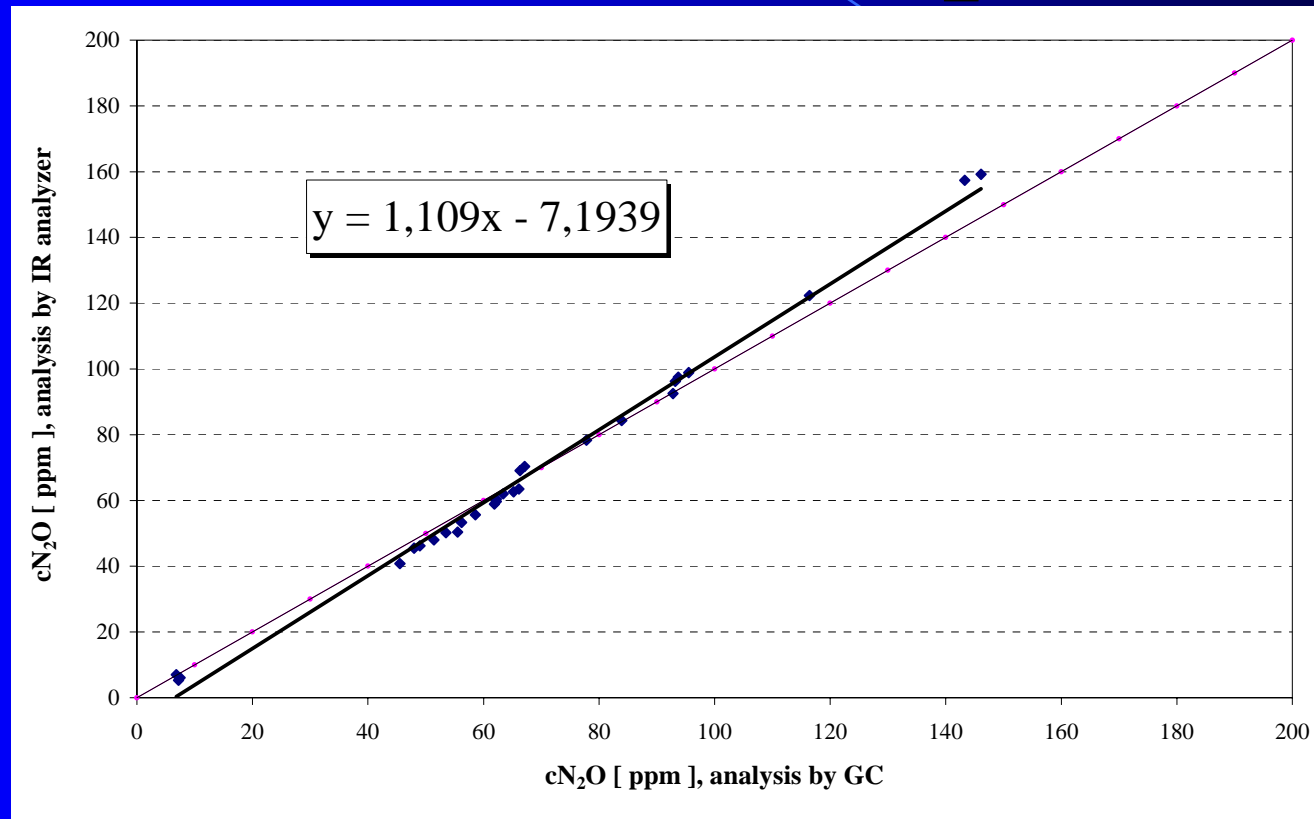
Experimental part

- IR spektrofotometry

- Advantage is continuous measurement
- Possibility of interferences CO_2 , C_xH_y , CO



Comparison of the IR and GC analyses of N₂O



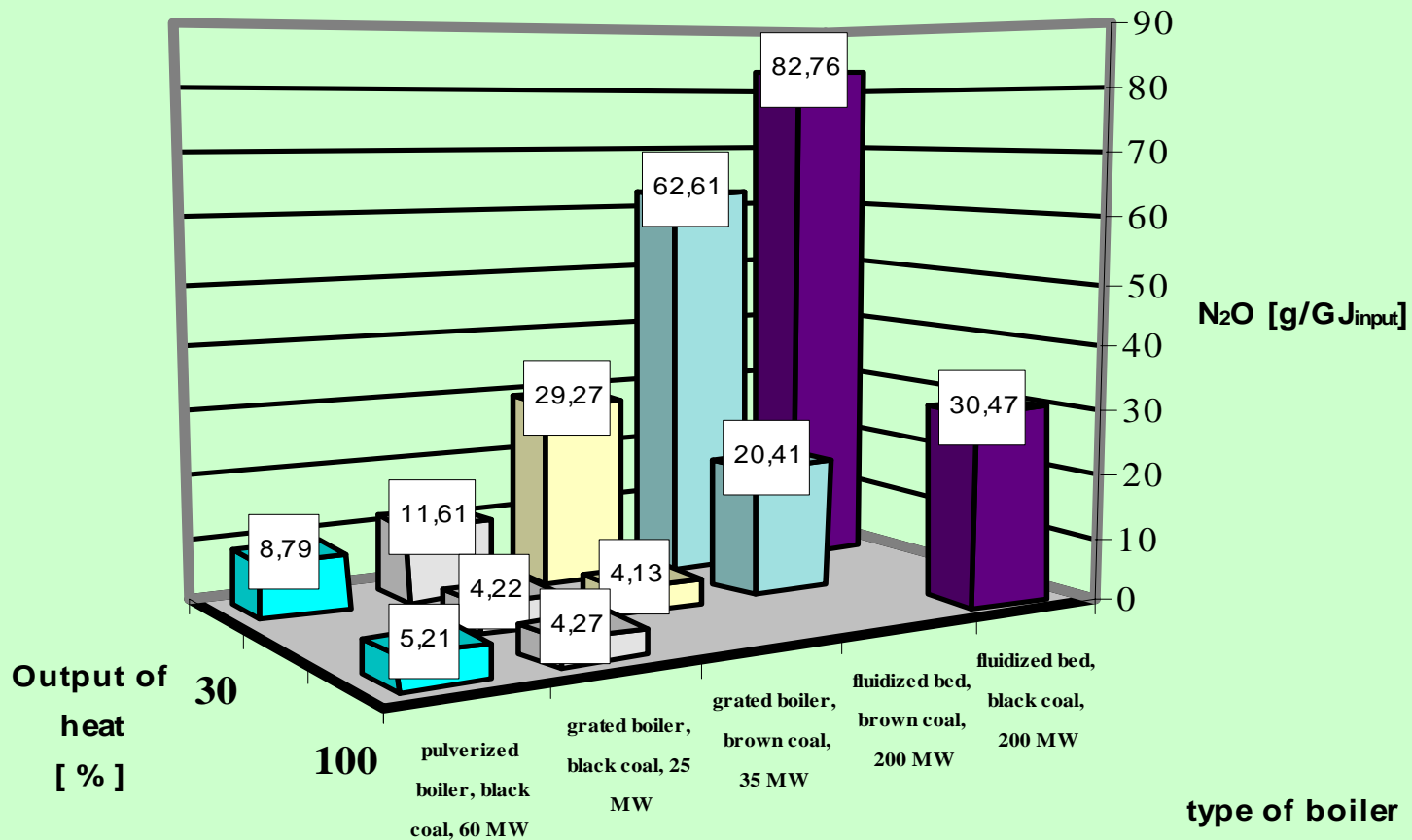
Maximal difference +/- 25 %, it is valid for content of CO when is lower than 500 ppm



Equipment for measurement



Coal combustion technologies, primary denitrification



Coal combustion technologies, secondary denitrification

SNCR – ammonia – reagent for deNO_x

Conversion of the reduced NO_x to N₂O, 3-8%,

SCNR – urea – conversion of the reduced NO_x to
N₂O is 2-2,5 higher than NH₃,

Emission factor, results from measurement in CZE	N ₂ O	NO _x
	g.GJ ⁻¹	g.GJ ⁻¹
Combustion proces with deNO _x	22,8	202,7
Combustion proces without deNO _x	4,2	296,3



Production of N₂O from municipal solid waste

- Noticeable source of nitrous oxide
- Very little work on their characterization
- Project researching on ERC Ostrava

N₂O formation by non-catalytic denitrification processes concerning energy utilisation of Waste



Production of N_2O from municipal solid waste

Municipal solid waste

Content of N – 0,1-1 mass %

*Gutierrez, Baxter,
Hunter, Svoboda

SNCR – **urea** – reagent for $deNO_x$

Conversion of the reduced NO_x to N_2O , 20-30%*,

SNCR – **ammonia** – reagent for $deNO_x$

Conversion of the reduced NO_x to N_2O , 10-15%*,

SCR – type of catalyst,

Conversion of the reduced NO_x to N_2O , 3-8%*,



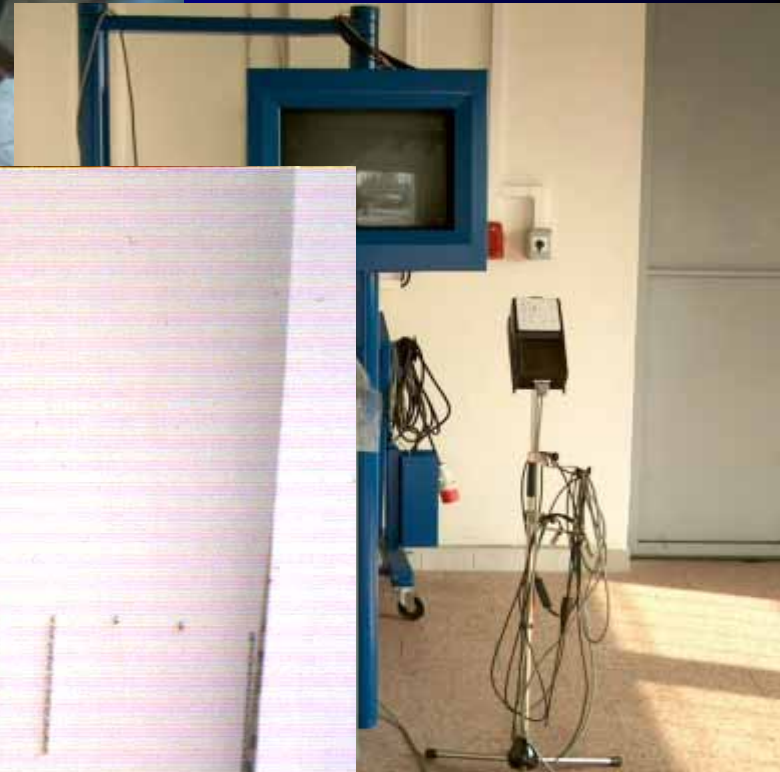
Production of N₂O from mobile sources

Mobile sources - cars

The following criteria of the measured cars were monitored with regard to the concentration level of nitrous oxide:

- Type of engine
- Type of fuel
- Operation output – revolutions of engine
- **With or without catalytic converter**
- **Age of catalytic converter**



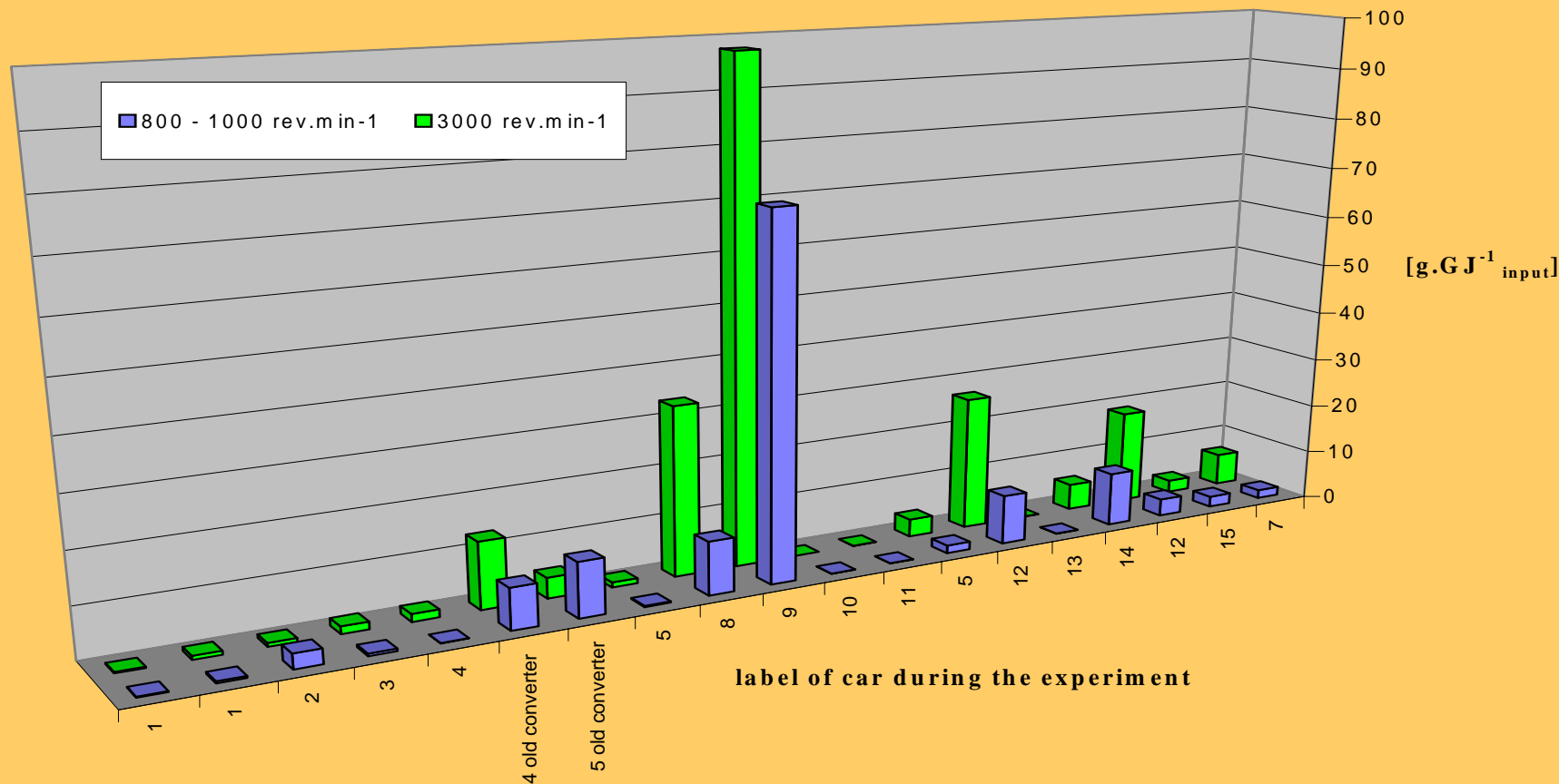


Catalyst for CO oxidation



Amount of CO in emission from cars is higher than 0,05%

Emission factors



1, 2, 3, 8, 9

engine without catalytic converter

4

engine with one-way catalytic converter

5, 10-14

engine with three-way catalytic converter

7, 15

diesel

12

LPG



Dependence on the age of catalytic converter

Engine with one-way converter	After running 5 000 km	After running 25 000 km
	N ₂ O [g.GJ ⁻¹]	
900 rev.min ⁻¹	0,1	17,2
3000 rev.min ⁻¹	3,1	24,7



Conclusion

- Denitrification of emissions from combustion processes in energy and transport increases emission of nitrous oxide
- There are general lack of results
- EF of NO_x decrease, new denitrification units will be installed and production of N_2O will be increased ?
- The work was performed thanks to the project No. **101/05/P278** of the Grant Agency of the Czech Republic.

