

# Ammonia and the *Secretum Secretorum*

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CEH, ECN, JRC

Commemorating the dual 250<sup>th</sup> Anniversary  
1756-2006

**Joseph Black:** demonstrated gaseous ammonia, Edinburgh

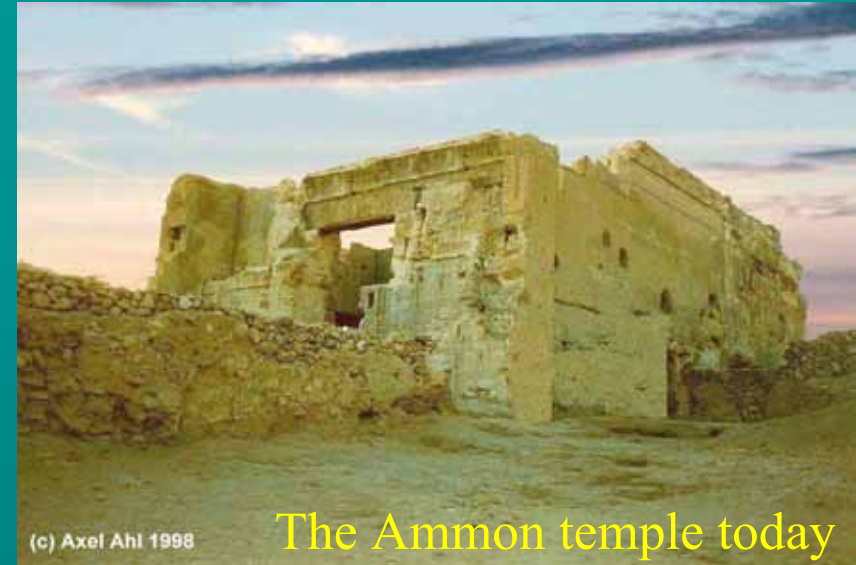
**James Davie & James Hutton:** established the first  
European *sal ammoniac* factory, Edinburgh

# **A fast history of ammonia....**

- >1000 years in 30 minutes
- From the present then working backwards
- Origin of scientific knowledge on  $\text{NH}_x$ ?
- Reflect on the messages from history for current  $\text{NH}_x$  research

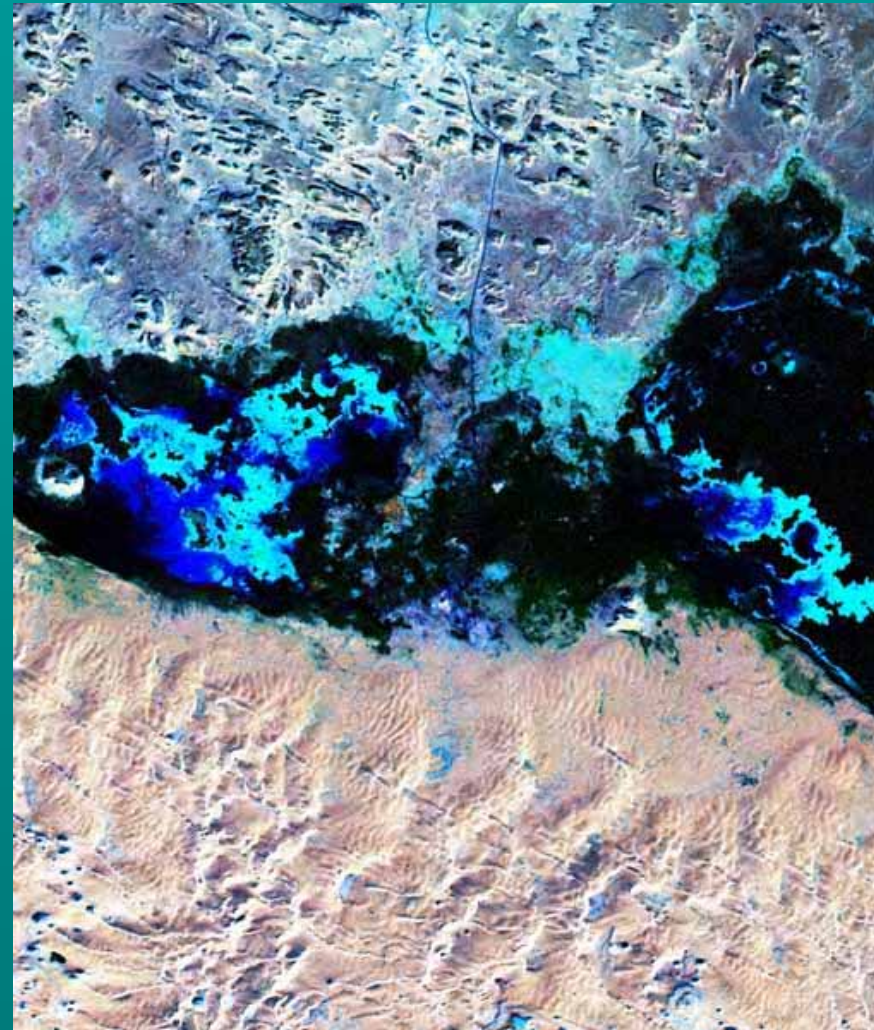
# The origin of the name “Ammonia”

- “Ammonium” a town at the Siwa Oasis, Lybia:
- “Temple of Ammon”: one of the three great oracles of the ancient world
- Ἀμμος = Greek for sand
- Ammon = Ram god, Amun of the sands.
- Major temple with over 80 priests; Oracle consulted by Alexander the Great (331 BC)



# Origin of *Sal Ammoniac*: Option A

A. Deliquescent salt found in the sands, used in Ammon sacrifices and prized in medicine (Pliny, Arrian, Columella etc)



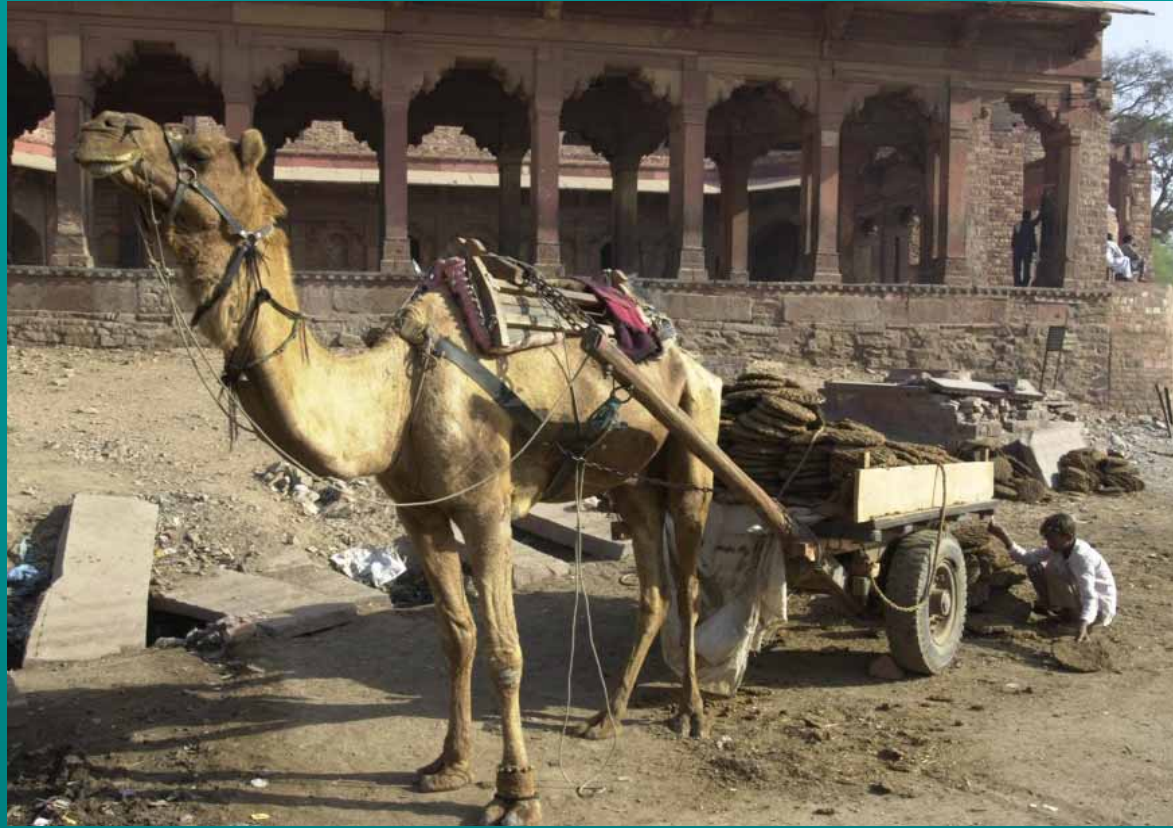
*Sal ammoniac* =  $\text{NH}_4\text{Cl}$

# Origin of *Sal Ammoniac*: Option B

B. Extract from the plant  
“Ammoniac”: *Dorema ammoniacum*  
Used in the temple ceremonies  
Pliny: “The price .. of the best quality  
is forty asses per pound”



# Origin of *Sal Ammoniac*: Option C



C. The priests found in soot on the temple walls in the soot from their burned sacrifices

# *Sal ammoniac* at ancient Ammonium.

## The Options

A. Pliny's salt from the sands  
by the Ammonium Oasis

B. Extract from the plant  
*Dorema ammoniacum*

C. On the sooty walls of the  
Temple of Ammon

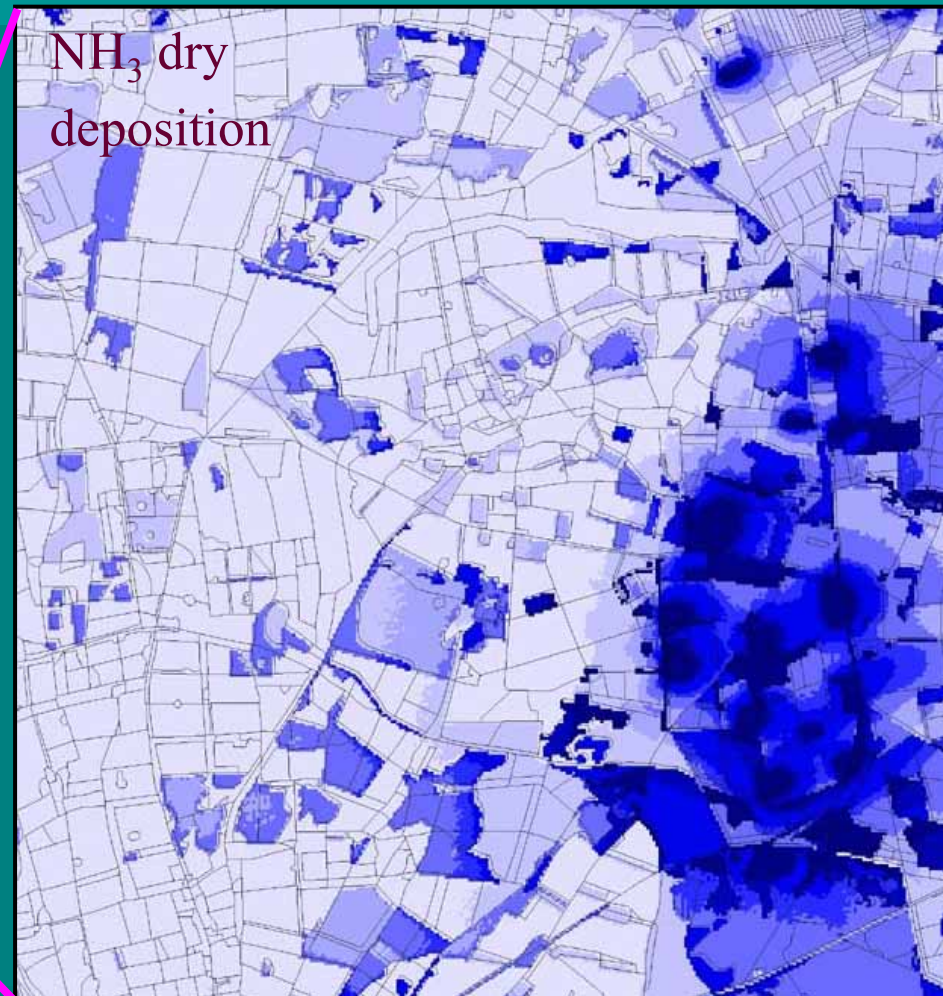
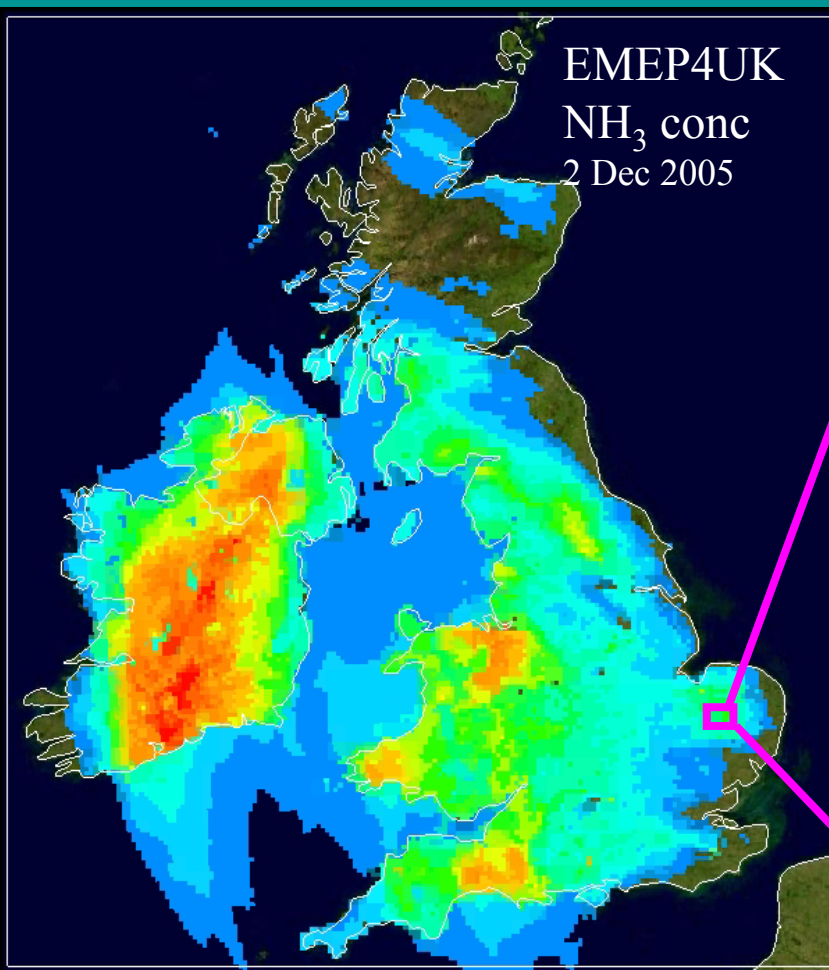
Your  
choice!

# Recent Achievements: The future history

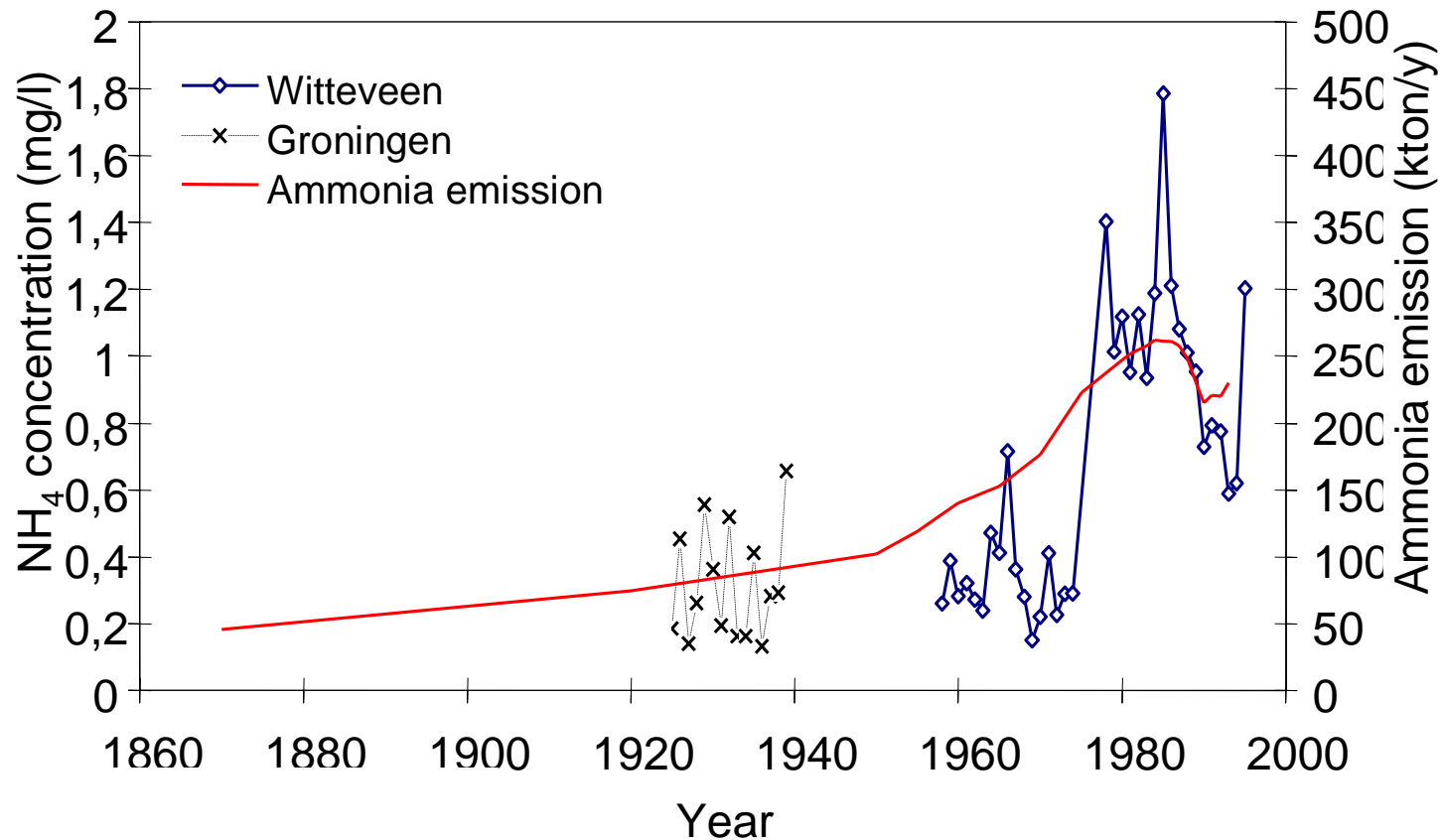
- What were the key  $\text{NH}_x$  advances of the last 25-30 years?
- **Examples:**
  - Atmospheric ammonia contributing to acidification and biodiversity loss.
  - Continuous flux measurements and understanding of bi-directional  $\text{NH}_3$  exchange
  - Massive data collection leading to comprehensive spatial  $\text{NH}_3$  emissions inventories
  - Numerical atmospheric transport models able to simulate atmospheric  $\text{NH}_x$  and chemistry
- **Questionnaire:** complete by Tuesday AM



# Multi-scale modelling of $\text{NH}_x$



# Ammonia emission and precipitation ammonium in the Netherlands



# Change in Agriculture after 1950

Nutrient shortage  
& priority to conserve



Fertilizer



Increased stock



Intensive livestock  
breeding

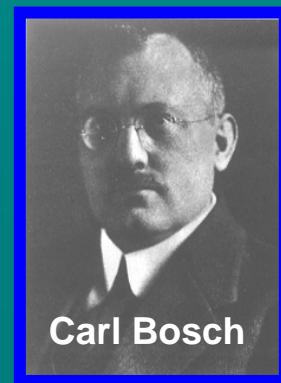
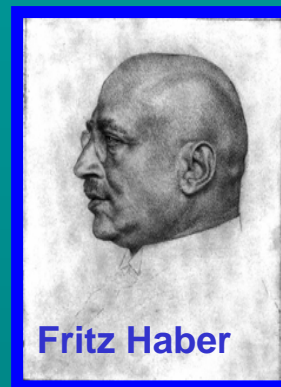
More  $\text{NH}_3$   
emission

# Haber-Bosch Process

- Limited supply of natural nitrate sources
- Haber-Bosch: 200 atmos; 450-500 °C



- Patented 1908 by Fritz Haber
- Commercialized 1910 by Carl Bosch
- Used Ostwald process (1902) to make  $\text{HNO}_3$  for  $\text{NH}_4\text{NO}_3$  production
- For explosives and fertilizers



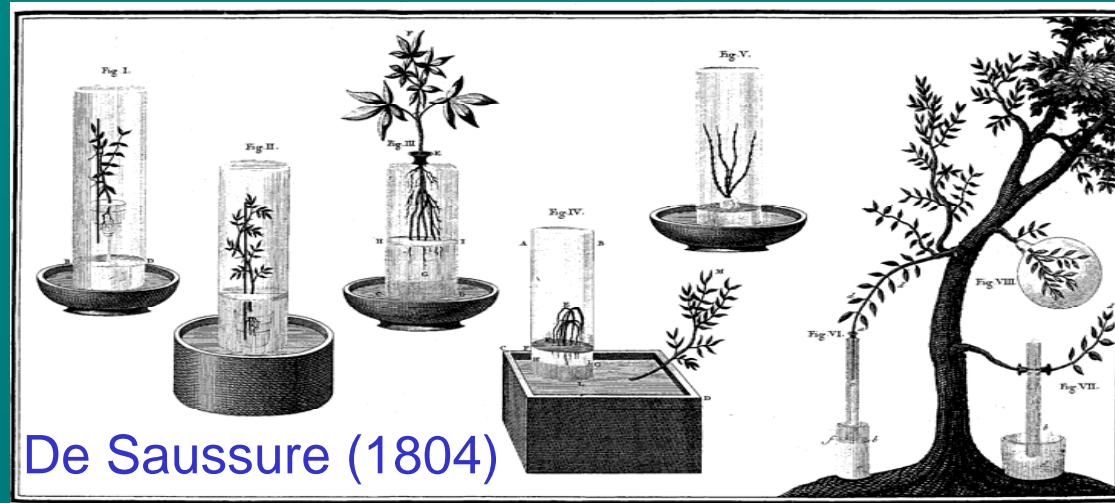
# Early 20<sup>th</sup> Century

- Role of atmospheric  $\text{NH}_x$  as a plant nutrient for agriculture
- Understood well the possibility for ammonia emissions and deposition
- First profile measurements of “dry deposition”, Rothamsted (Hall & Miller 1911)
- $\text{kg NH}_x \text{ ha}^{-1} \text{ yr}^{-1}$  to covered plate collectors:

Height of collector	Broadbalk (Fertilized)	Lawn (unfertilized)
1.15 m	1.28	1.77
0.05 m	1.98	1.02

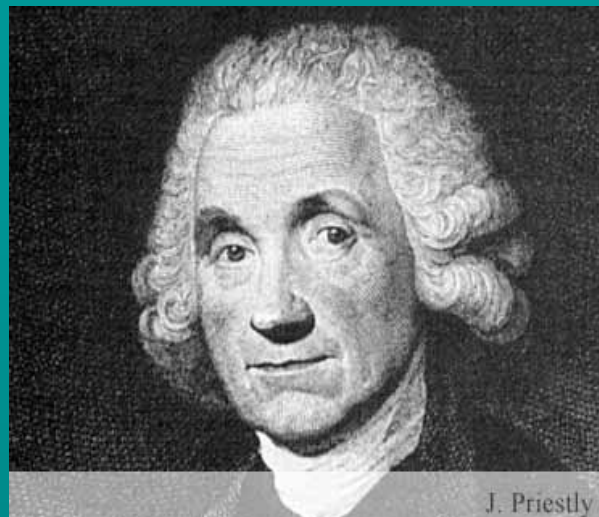
# 19<sup>th</sup> Century debate on the source of nitrogen for plants

- **De Saussure** (1804): Early studies of plant N nutrition. Demonstrated the presence of  $\text{NH}_3$  air
- **Liebig**: Argued that plant N would be obtained from atmospheric  $\text{NH}_x$ , like  $\text{CO}_2$ .
- **Lawes & Gilbert, and Way** at Rothamsted showed that inputs of wet deposition were insufficient for crop growth. Demonstrating the importance of soil N.
- **Ville, Schlosing etc.**  
1850-1900: First atmos.  $\text{NH}_3$  conc. measurements across Europe.  
Highest values in cities



# 18<sup>th</sup> Century Discovery of Ammonia in Air

- **Joseph Black (1756)** : Thesis University of Edinburgh. Demonstration of the volatile alkali on adding lime to *sal ammoniac*
- **Joseph Priestly (1774)**: Captured the gas liberated above heated spirit of hartshorn in a trough above mercury “alkaline air”, reacted it with HCl and made *sal ammoniac*
- **Karl Wilhelm Scheele (1777)** Showed alkaline air contains nitrogen
- **Morveau / Bergmann (1782)** Named it “ammonia”
- **Claude-Louis Berthollet (1785)**: Showed that the chemical composition was 0.81 g N : 0.19 g H.



J. Priestly



Berthollet

## 18<sup>th</sup> Century Debate on the source of *sal ammoniac*

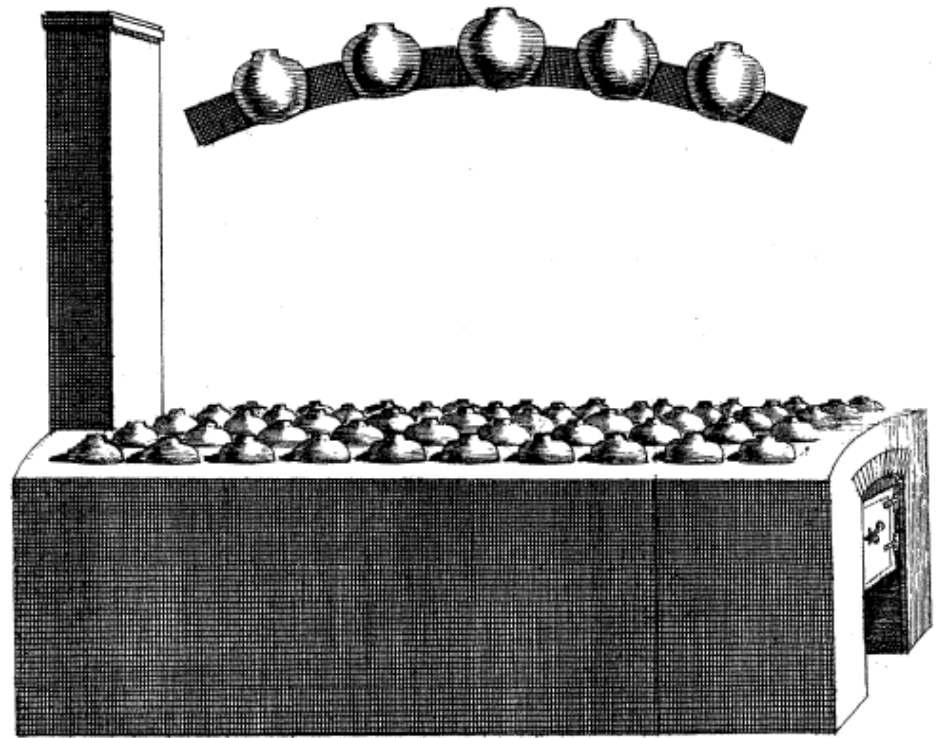
- Long running debate at the Academie Francais
- Imported from Egypt since 12<sup>th</sup>C, but the process unknown and kept secret by Venetian traders.
- Pliny's description of *sal hammoniacum* doubted.
- Geber (13<sup>th</sup> C) & Glauber (1567) claimed by boiling down urine, NaCl & soot (5:1:½). Later shown this was not effective: was it deliberate mis-information?
- Siccard (Academie Francais) and later Lineaus reported to Royal Society (1760) the detailed process.



# Lineaus, Hasselquist & Ellis on: Egyptian *Sal ammoniac* manufacture (1760)

- Soot from Spring dung (from legume grazing) was collected from peasants & placed in jars
- Jars gradually heated above a furnace until it reached “hell fire” ( $>300\text{ }^{\circ}\text{C}$ )
- Sublimed *sal ammoniac* collected in necks of glass vessels

*Philos. Trans. Vol. LI. TAB. XI. p. 505.*



*A Plan of the subliming Furnace or Oven for making Sal Ammoniac in Egypt  
taken from the description here given.*

# The first European *sal ammoniac* factory

James Davie & James Hutton  
(1756), Edinburgh

- Secrecy over the process
- Appears they broadly followed the Egyptian approach
- They used the soot of coal fires
- Took ALL the soot of the *Tronmen* society of Edinburgh sweeps
- 26 kg of soot yielded 6 kg of *sal ammoniac*



19<sup>th</sup>C Edinburgh from the NW

From Session Papers in the  
Library of Writers to the Signet

# Ammonia in European alchemy

- **Arrival in Europe:** *Sal ammoniac* known from ~1140 AD  
Latin translations of Arabic alchemy texts
- **Four spirits of medieval alchemy:** mercury, sulphur, arsenic and *sal ammoniac*
- **Spirit of Hartshorn:**  $\text{NH}_4\text{CO}_3$  from distillation of horn, bone, offal etc
- **The key to *Aqua Regia*:**  $\text{NH}_4\text{Cl} + \text{HNO}_3$  dissolves gold.  
First step to making gold...
- **Fulminating Gold:**  $\text{AuNH.NH}_2$ . (attrib. Valentinus 15<sup>th</sup> C)  
Precipitate out  $\text{AuOH}$ , then add Spirit of Hartshorn.  
The most expensive explosive.  
Used in Battle: Dutchman for English vs French in 1628!
- **The Emerald Tablet** and the search for the elixir, the philosophers stone....

# Isaac Newton and the Emerald Tablet

- Newton also an alchemist and translated the *Emerald Tablet* - mystic text of 12 lines.

*7a. Seperate thou ye earth from ye fire, ye subtile from the gross sweetly wth great indoustry.*

*8. It ascends from ye earth to ye heaven & again it desends to ye earth and receives ye force of things superior & inferior.*

- It was believed to contain the full secret of Medieval alchemy – to the initiated...
- Appeared the *Secretum Secretorum* of Roger Bacon (1255). and attrib. to *Hermes* (Egyptian 400 BC?)

# Newton's experiments with alchemy ~1680

I sublimed  $\text{♁}$  &  $\text{♁}$  ana.  
 Of this sublimate I  
 took an ounce & 1/2  
 an ounce of  $\text{♁}$  tiate  
 & sublimed them.

.. there arose a little salt  
 like  $\text{♁}$  which ... had  
 a sweetish tast.

$\text{♁}$  Sal ammoniac

$\text{♁}$  ana Antimony ore

$\text{♁}$  tiate Lead antimonate

13. I sublimed y<sup>e</sup> grossest  $\text{♁}$  <sup>subl.</sup> 9 parts from A viz 1 part that  
 is 180 gr of  $\text{♁}$  from 20 gr of  $\text{♁}$  there was a light  
 black sooty matter 24 gr <sup>of  $\text{♁}$  from shape much</sup> this I added to white  $\text{♁}$  72 gr. &  
 from ~~600 gr~~ it sublimed 600 gr of fresh  $\text{♁}$  with a  
 stronger & longer heat there remained 90 gr wch I put  
 in a firshovel & held it over y<sup>e</sup> fire till it had done  
 fuming & then it weighed 86 gr or 87 gr. The sublimate  
 was much whiter than y<sup>e</sup> first. ~~being only a fally light white~~ I secured  
 when I put but a little into water to dissolve almost  
 all of it. But upon dissolving it all ~~there~~ the calx  
 wch fell being dryd weighed 170 gr.  
 I dissolved  $\text{♁}$  in  $\text{A}$  4 parts &  $\text{X}$  1 part & precipi-  
 tated the solution wth water & edulcorated & dried y<sup>e</sup>  
 precipitates & sublimed  $\frac{3}{4}$  wth  $\frac{1}{2}$  of  $\text{X}$ . ~~there~~  
~~of  $\text{♁}$~~  there remained in y<sup>e</sup> bottom  $\frac{1}{8}$  ~~of  $\text{♁}$~~   
 of light white calx wch sublimed again wth  $\frac{1}{4}$  of  $\text{X}$   
 left  $\frac{1}{12}$  of white calx in y<sup>e</sup> bottom. The first of  
 these two sublimate 12 gr I mixed wth calx of salami  
~~alright~~ ground them fine together & put them on a glas  
 in the fire. After y<sup>e</sup> sublimate was most of it fumed  
 away the residue melted & boyled a while & when  
 away the residue in a red heat I weighed it a

# Arab origin of Alchemy and ammonia

## Convention:

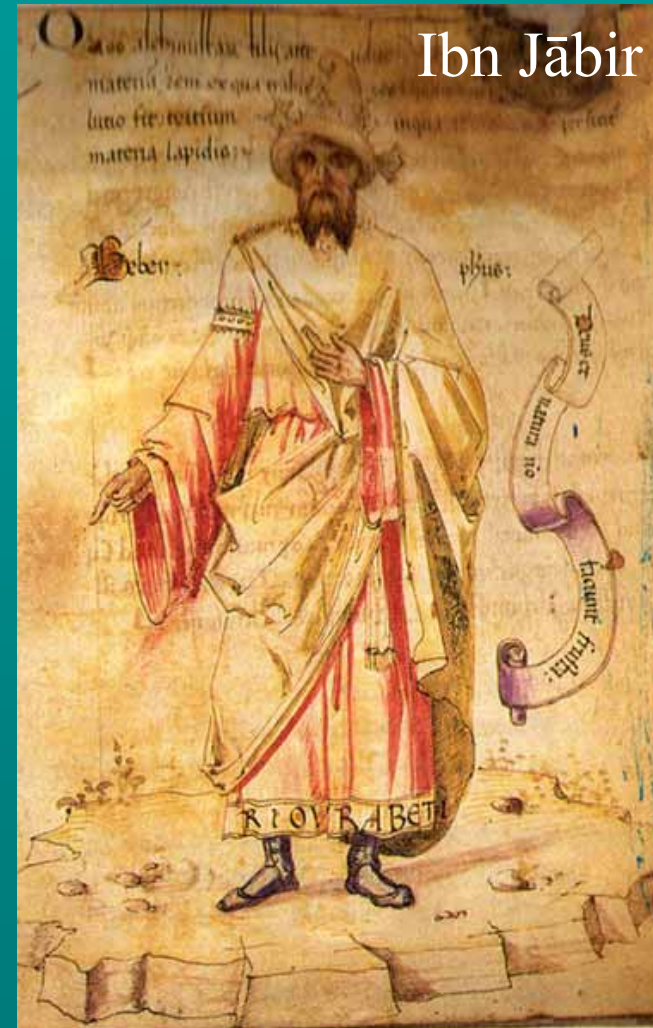
- Discovered  $\text{NH}_x$  in 9<sup>th</sup> C Iraq
- **Ibn Jābir**: *sal ammoniac*, *Aqua Regia*
- **Al Razi**: Spirit of hartshorn

## Current understanding

- Jābirian Corpus – by many authors
- Exploited  $\text{NH}_x$  but not discovered.
- Earliest copy of *Kitāb Sirr al-Asrār*
- From earlier texts: Trans-Oxiana, China?

## Uses of *sal ammoniac*

- Colour and dissolve metals
- Reduce metal oxides, for tinning etc.
- Dying cloth, glass pigments
- Medical: chest, Hg antidote etc



# Sources of $\text{NH}_x$ in Arab times

## (texts from c. 930-970 AD)

### Sal ammoniac ( $\text{NH}_4\text{Cl}$ )

- **Al-Mas'ūdī, Al-Muqaddasi:** Naturally burning coal seams in Transoxiana and China
- **Ibn Hauqal:** Damindan volcano, SE Persia (Others called it “a smoking fountain of hell”)
- **Al-Maquaddasi:** Sicily – but the supply already exhausted
- **Al- Istakhrī:** From Egypt, by sublimed soot of burned camel dung at the Baths

### Spirit of Hartshorn ( $\text{NH}_4\text{CO}_3$ )

- **Al-Razi and others:** Distillation of Hair and offal for  $\text{NH}_4\text{CO}_3$ : New in Arabia?

# Changing names of $\text{NH}_4\text{Cl}$

## Major variants only!

- Ammonium chloride (English, 19<sup>th</sup>-21<sup>st</sup> C)
- Sal ammoniac (Latin, 12<sup>th</sup> -19<sup>th</sup> C) (Egyptian source)
- Sal armoniac (Latin, 10<sup>th</sup>-15<sup>th</sup>. C) (Armenian source)
- Almisader, Mizadir (Latin, 12<sup>th</sup>-15<sup>th</sup>C)
- Nushadir (Arabic, 9<sup>th</sup>-present)
- Nao Sha (China, 1<sup>st</sup> C- present)
- Nao (China, origin unclear)

## Possible etymology

Nao = impure Nao Sha (J. Needham FRS FBA, 1980 favoured)

Sha = sand/granules

Dur = medicine? (Stapleton, 1905)



# Chinese origin *sal ammoniac*?

**Wei-Po Wang:** *Tshan Thung Chhi* (142 AD)

*“It would be like mending a cauldron with glue, or bathing a boil with sal ammoniac, or driving away cold with ice....”*

BUT

**Su Jing:** *Tang Ben Cao* (658-659 AD)

*“Nao Sha originated from tribesmen at the western border of the Zhou dynasty”* (1000-200 BC)

- Tarim Basin (Xing Jiang) the source of burning coal fields
- Tarim was Iranian speaking, e.g. Sogdians
- Movement of culture was from the west: Zoroastrianism, Buddhism, Manicheism, Nestorian Christianity (500 BC – 500 AD)
- Iranian Magi (Zoroastrians) as priests at the Chinese Zhou court (1000-200 BC)



# Western peoples in ancient Tarim



Mummy from Zaghunluq  
Tarim Basin, (~1000 BC)

# Zoroastrian & Mithraic origins of sal ammoniac

- Bactria, Fergana, Sogdia: popn. 1 million
- The *other* fertile crescent (from 3000 BC)
- Eternal coal fires at several locations
- Mithraic & later Zoroastrian Fire ‘worship’
- NushAdar: hero, 4<sup>th</sup> generation after Zoroaster (~400-500 BC?)

## Etymology:

- Nush = **Immortal**; Adar = **Fire**

later variant:

- Nush = Elixir or Antidote



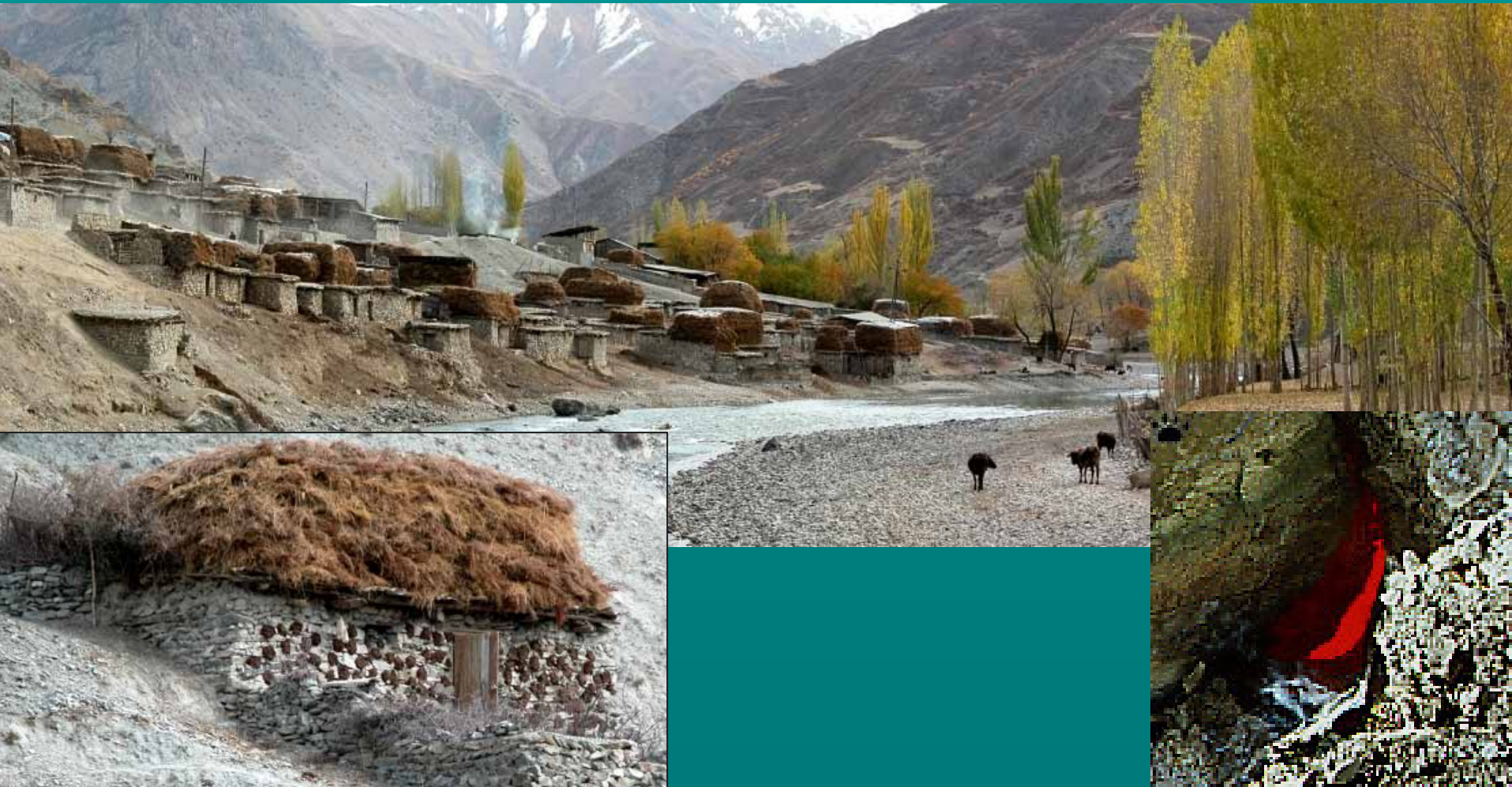
Ahura Mazda & sacred fire



Fire temple, Perseopolis

# Anzob, upper Zerefshan (Tajikistan)

- Burning coal caves visited by Alexander
- Rich deposits of Au, Hg, Ag, Sb etc.



# Assyrian Evidence of *Sal ammoniac* (~500 BC)

## In pigments?

- To make Lead Antimonate  $\text{Pb}_3(\text{SbO}_4)_2$  (Naples Yellow) used in glazed tiles

## In Sumerian Cuniform texts?

- IM.KAL.LA = sublimate of soot = *sal ammoniac* (Thompson, 1933)



Ahura Mazda & Sphinxes



# Zozimus (Egypt, 300 AD) on the Elixir

- Alchemist, gives his interpretation of Agathodaimon (? BC),
- Noting the properties of the philosopher's stone he writes:

*This is the uncommunicated mystery which none of the prophets dared to divulge in words but revealed only to the initiated.*

*In their symbolical scriptures they called it the stone which was not a stone, the thing unknown and yet known to everyone, the despised thing of great price, the thing given by God and yet not given.*

*For my part I shall praise it... for it is the one thing which dominates matter.*

*Such is the drug of power, the Mithraic mystery.*

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*Such is the **drug of power, the Mithraic mystery.***

# Conclusions

## *Sal Ammoniac* at the Temple of Ammon?

- A. Pliny's salt in the sand? It was a prized NaCl. Europe looked for parallels in classical writers. Or a traders trick.
- B. The plant extract? *Gum Ammoniac* came from the town of Ammonium, but no  $\text{NH}_x$  in it. The red herring.
- C. Crystals in the Temple Soot? 18<sup>th</sup>C Historical extrapolation...

But  $\text{NH}_4\text{Cl}$  from soot sublimation in Egypt was:

A major industry in ~950 AD

Probably known by Zosimus (300 AD)

Possibly known by Agathodiamon (? BC)



# Spread & sal ammoniac along the Silk Road



# Conclusions and our $\text{NH}_x$ research

Tend to forget the understanding of previous cultures  
The recollection sets perspective.

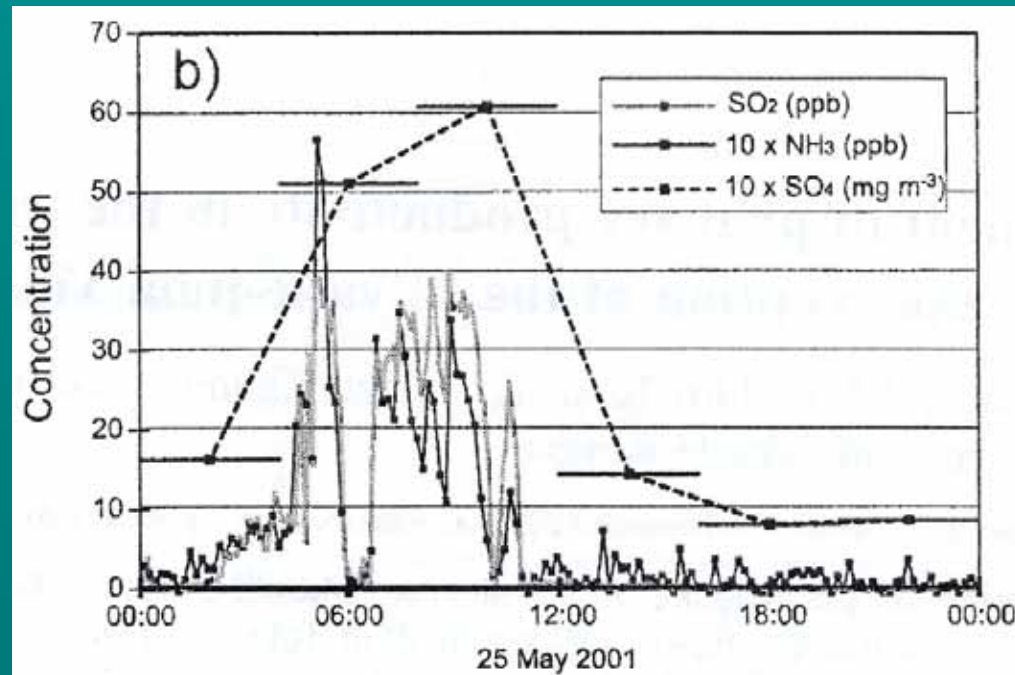
## The original sources of $\text{NH}_3$ are hardly studied:

- $\text{NH}_3$  from Dung burning (no literature)
- Direct  $\text{NH}_4\text{Cl}$  from all biomass burning (missing)
- $\text{NH}_3$  &  $\text{NH}_4\text{Cl}$  from coal combustion (minimal data)
- $\text{NH}_3$  and  $\text{NH}_4\text{Cl}$  from volcanoes.....

# Volcanic ammonia & sal ammoniac

- Le volcan de la Soufrière, Guadeloupe “Many fumeroles give off an unpleasant odour of rotten eggs & ammonia”
- Uematsu et al. (2004) *Geophys Res Lett.*

Mijake-jima emission:  
300 kt  $\text{NH}_3$  per year!



100 km from Mijake-jima, Japan